



Terms and phrases to be used:

Decomposition Reaction

Decrease of temperature

Double Displacement Reaction

Endothermic Reaction

Exothermic Reaction

General Equation: $A + B \rightarrow AB$

General Equation: $A + BC \rightarrow AC + B$

General Equation: $AB \rightarrow A + B$

General Equation: $AB + CD \rightarrow AD + CB$

Heat is absorbed by the reaction.

Heat is released from the reaction.

Increase of temperature

One element changes places in a compound with another element.

One substance breaks down into smaller substances.

Single Displacement Reaction

Synthesis Reaction

Two compounds switch elements within a chemical reaction.

Two substances combine to form one new substance.