

Name \_\_\_\_\_

# Elements

Read each element, characteristic, or description. Decide which category that description best fits into and write its number in the appropriate section of the box below.

- |   |   |
|---|---|
| 1. Shiny                                  | 6. Iron                                   |
| 2. Poor thermal and electrical conductors | 7. Good thermal and electrical conductors |
| 3. Ductile                                | 8. Could be malleable or unmalleable      |
| 4. Brittle                                | 9. Oxygen                                 |
| 5. Could be shiny or dull                 | 10. Sometimes called semiconductors       |

Metals	Nonmetals	Metalloids

Use the words from the box below to complete the following paragraph about the periodic table.

nucleus	protons	mass	atomic number
below	location	neutrons	element

Each square on the periodic table includes lots of information about an \_\_\_\_\_.

11

The one-letter or two-letter abbreviation tells the element's name. The number above the abbreviation

is the \_\_\_\_\_.

12

This number is the number of \_\_\_\_\_ in each atom of

13

the element. The number \_\_\_\_\_ the chemical abbreviation is the atomic

14

\_\_\_\_\_ number. This number is the number of protons plus the number of

15

\_\_\_\_\_ in the \_\_\_\_\_ of the atom. The \_\_\_\_\_ of the

16

17

18

element on the table tells what type of element it is based on its properties.

Look at the information in the box below, then fill in the blanks.

- Element: \_\_\_\_\_
- Symbol: \_\_\_\_\_
- Atomic Number: \_\_\_\_\_
- Atomic Mass: \_\_\_\_\_
- Number of Protons: \_\_\_\_\_
- Number of Neutrons: \_\_\_\_\_

7
N
14.0