Name: lonic and Covalent Bonding Concept Map Two main types of bonding Type of elements Type of elements in this bond in this bond bond is formed by bond is formed by (what happens with e-?) (what happens with e-?) Covalent bonds form Ionic bonds form an example is Properties include: properties include: an example is

Soluble in water

Low boiling point  $\checkmark$ 

lonic /

Sodium chloride, NaCl V

Nonmetals /

Crystalline compounds  $\sqrt{\phantom{a}}$ 

Low melting point /

High melting point  $\checkmark$ 

Two main types of Bonding/

Not soluble in water

Metals v

Conductive aqueous solutions

Molecular compounds  $\checkmark$ 

Electrons are shared ~

Not conductive  $\checkmark$ 

Covalent /

High boiling point  $\sqrt{\phantom{a}}$ 

Electrons are transferred /

Ammonia NH<sub>3</sub>/