

## The Polyatomic Ions

### Minus One Charge

$\text{H}_2\text{PO}_4$	Dihydrogen Phosphate
$\text{HSO}_3$	Hydrogen Sulfite
$\text{HSO}_4$	Hydrogen Sulfate
$\text{HCO}_3$	Hydrogen Carbonate
$\text{NO}_2$	Nitrite
$\text{NO}_3$	Nitrate
$\text{ClO}$	Hypochlorite
$\text{ClO}_2$	Chlorite
$\text{ClO}_3$	Chlorate
$\text{ClO}_4$	Perchlorate
$\text{C}_2\text{H}_3\text{O}_2$	Acetate
$\text{CN}$	Cyanide
$\text{OH}$	Hydroxide
$\text{MnO}_4$	Permanganate

### Minus Two Charge

$\text{HPO}_4$	Hydrogen Phosphate
$\text{SO}_3$	Sulfite
$\text{SO}_4$	Sulfate
$\text{CO}_3$	Carbonate
$\text{CrO}_4$	Chromate
$\text{Cr}_2\text{O}_7$	Dichromate
$\text{C}_2\text{O}_4$	Oxalate
$\text{SiO}_3$	Silicate

### Minus Three Charge

$\text{PO}_4$	Phosphate
$\text{PO}_3$	Phosphite

### Plus One Charge

$\text{NH}_4$	Ammonium
---------------	----------