- Notes:
 - <u>http://www.usca.edu/chemistry/genchem/sigfig.h</u>
 <u>tm</u>
 - <u>http://www.purplemath.com/modules/rounding2</u> .<u>htm</u>
- Practice:
 - <u>http://www.chem4free.info/calculators/signdig.ht</u>
 m

Unit 1: Scientific Notation

Significant Figures

 Digits that carry meaning contributing to its precision.

All Numbers Except

- All leading zeros;
- Trailing zeros when they are merely placeholders to indicate the scale of the number; and
- Digits that cause a solution to have a greater precision than the equipment supports.

- All non-zero digits are considered significant.
- For example:
 - 91 has two significant figures
 - 123.45 has five significant figures

Rules

- Zeros appearing anywhere between two non-zero digits are significant.
- For example:
 - 101.1203 has seven significant figures

Rules

- Leading zeros are not significant.
- For example:
 - 0.00052 has two significant figures

Rules

- The significance of trailing zeros in a number not containing a decimal point can be ambiguous.
- A decimal point may be placed after the number.
- For example:
 - "100." has three significant figures